Semester V								
S.No	Course Code	Course Name	L	Т	P	С		
1	CH 323	Coordination and organometallic chemistry	3	0	0	6		
2	CH 321	Concepts in organic synthesis	3	0	0	6		
3	CH 322	Molecular spectroscopy	3	0	0	6		
4	CH 312	Chemistry laboratory-II	0	0	3	3		
5		Program Elective-II	2	1	0	6		
6		Program Elective-III	2	1	0	3		
7		Program Elective-IV	2	1	0	3		
		Total Credits				33		

1	Title of the course	Coordination and organometallic chemistry			
1	(L-T-P-C)	(3-0-0-6)			
2	Pre-requisite courses(s)	Fundamental concepts and applications of chemistry (CH101) and Transitional metals and coordination chemistry.			
3	Course content	Coordination compounds: Valence bond theory, crystal field theory, molecular orbital theory and their applications, inner sphere electron transfer, outer sphere electron transfer, classification of ligands, trans effect, stability constant, Jahn-Teller effect, ploy nuclear complexes, reaction of coordination compounds. Bonding and Electronic Spectra: MO theory of transition metal complexes spectroscopic term symbols, selection rules, Orgel diagrams, and charge transfer bands; Magnetism of Coordination Complexes. Organometallic chemistry: General concepts: Types of ligands, soft vs hard ligands. 18e rule and its exceptions, isolobal and isoelectronic analogies. s and π bonding, Structure, bonding and reactivity studies of metal carbonyls, nitrosyls, dinitrogen complexes. Organometallic Reactions and Mechanisms: oxidative addition, reductive elimination reactions, organometallic complexes with metal-metal bonds. Metal-ligand Multiple Bonds: Fischer and Schrock type carbene complexes, carbyne complexes, and metal-heteroatom (O/N) multiple bonds			
4	Texts/References	 R. H. Crabtree, The Organometallic Chemistry of the Transition Metals, 6ed, Wiley, 2013. J. Hartwig, Organo-transition Metal Chemistry: From Bonding to Catalysis, University Science Books, 2010. B. D. Gupta and A. J. Elias, Basic Organometallic Chemistry: Concepts, Syntheses and Applications, 2ed, Universities Press, 2013. G. L. Miessler and D. A. Tarr, Inorganic Chemistry, 3ed, Pearson, 2008. B. Douglas, D. McDaniel, and J. Alexander, Concepts and Models of Inorganic Chemistry, 3ed, Wiley, 2010. J. E. Huheey, E. A. Keiter, and R. L. Keiter, Inorganic Chemistry Principles of Structure and Reactivity, 4ed, Pearson Education, 2006 Inorganic Chemistry. D. F. Shriver, and P. W. Atkins. 3rd Edn. Oxford University, Oxford, 1999. S. F. A. Kettle, Physical Inorganic Chemistry – A Coordination Chemistry Approach, Springer, 1996. 			

1	Title of the course Concepts in organic synthesis		
	(L-T-P-C)	(3-0-0-6)	
2	Pre-requisite courses(s)	Fundamental concepts and applications of chemistry (CH101) and organic reactions and reagents	
3	Course content	Synthesis, reactions, mechanisms, and selectivity involving the following-alkenes, alkynes, arenes, alcohols, phenols, aldehydes, ketones, carboxylic acids and their derivatives, halides, nitro compounds and amines. Use of compounds of Mg, Li, Cu, B and Si in organic synthesis. Concepts in multistep synthesis- retrosynthetic analysis, disconnections, synthons, synthetic equivalents, linear and convergent synthesis, reactivity umpolung, selectivity, protection and deprotection of functional groups, Asymmetric synthesis: Chiral auxiliaries, methods of asymmetric induction – substrate, reagent and catalyst-controlled reactions; enantiomeric and diastereomeric excess; enantio-discrimination. Resolution – optical and kinetic.	
4	Texts/References	 Clayden, J., Greeves, N., Warren, S., Wothers, S. Organic Chemistry, Oxford University Press, 2001. Carruthers, W., Coldham, I. Some Modern Methods of Organic Synthesis, Cambridge University Press, 2004. Smith, M. B. and March, J. Advanced Organic Chemistry, Wiley Interscience, 2007. Carey, F. A., Sundberg, R. J. Advanced Organic Chemistry, Part A and B, Springer, 2007. Smith, M. B. Organic Synthesis, McGraw-Hill, 2001. Warren, S. Organic Synthesis: The Disconnection Approach, Wiley, 1983. G. S. Zweifel and M. H. Nantz, Modern Organic Synthesis-An Introduction, W. H. Freeman and Company, 2006 	

1	Title of the course	Molecular spectroscopy		
	(L-T-P-C)	(3-0-0-6)		
2	Pre-requisite courses(s)	Fundamental concepts and applications of chemistry (CH101)		
3	Course content	Introduction to spectral energy domains and measurement of spectra, Implications of discrete energy levels, Population of States – Boltzman Distribution, Interaction of radiation with matter, origin of linewidths in molecular spectra, Transition dipole moment and Fermi's Golden Rule, Einsteins Coefficients, Lasers and Masers. Rotational (Microwave) spectroscopy, Molecular vibrations - Infrared spectroscopy, Normal mode analysis, Raman Scattering, Selection Rules from Group Theory, Molecular electronic spectra, Photophysical processes, Non-Linear Spectroscopy, Nuclear Magnetic Resonance, Relaxation times, FT-NMR, spin-spin coupling, ESR, Nuclear Quadrupolar Resonance.		
4	Texts/References	 J. L. McHale, Molecular Spectroscopy, Pearson Education, 1999. M. Hollas, Modern Spectroscopy, Wiley; 4th edition, 2004. F. A. Cotton, Chemical Applications of Group Theory, 3rd edition, Wiley-Interscience, 1990. D. C. Harris, M. D. Bertolucci, Symmetry and Spectroscopy, Dover, 1990. C. M. Banwell, E. M. McCash, Fundamentals of Molecular Spectroscopy, Tata McGraw Hill, 1983 G. M. Barrow, Molecular Spectroscopy, McGraw Hill, 1962 J. I. Steinfeld, Molecules and Radiation: An Introduction to Modern Molecular Spectroscopy, 2nd edition, Dover, 2005. J. D. Graybeal, Molecular Spectroscopy, McGraw Hill 1993. D. A. McQuarrie and J. D. Simon, Physical Chemistry - a molecular approach, Viva Books Pvt. Ltd. 1998. 		